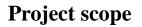


# An Introduction to Ammonia Spray Lab Safety and Setup

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- Fundamental ammonia spray breakup data from experiments
- Mixing data for ammonia-air/nitrogen mixtures
- Simulation of ammonia spray and validation against data from experiments

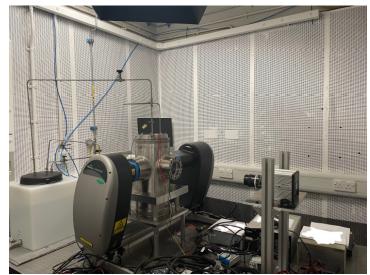
#### The current task – this presentation

- Build a lab from scratch for ammonia spray tests
- Undertaken safety precautions



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 $\uparrow Oxford$  Ammonia Spray Lab in May 2022



†Oxford Ammonia Spray Lab in June 2023

# **Ammonia supply**

#### Supply available

- Saturated liquid-vapour mixture
  - Filling ratio in accordance with the BCGA Code of Practice 35 [1] Ο
  - Cylinder pressure depends on ambient temperature [2] Ο
- Best supply available: Saturated liquid  $\geq$

#### **Requirements for ammonia spray tests:**

High injection pressure

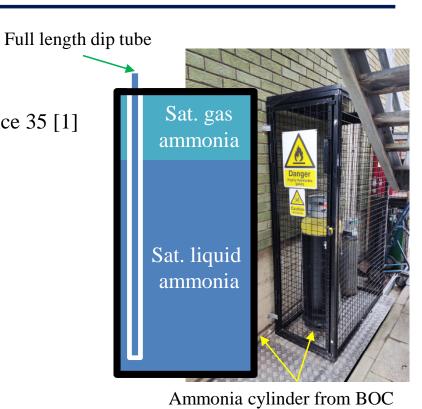
Pressure (bar)

- Liquid injection without cavitation ۰
- Need an external pressurising system

\*Locations of the markers in this plot are for illustration purposes only.

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Stull, 1947 Liquid



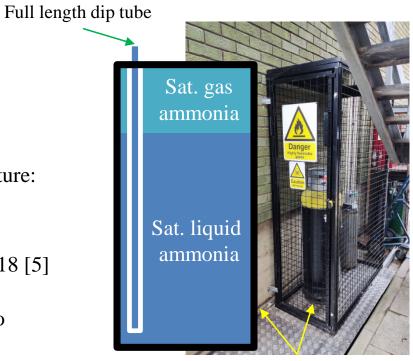
#### 60 50 40 Vapour 30 40 °C, 15.5 bar 20 -5 °C, 3.5 bar 10 Plot adapted from [2] 180 200 220 240 260 280 300 320 340 360 Temperature (K)



# **Toxicity considerations and ammonia storage**

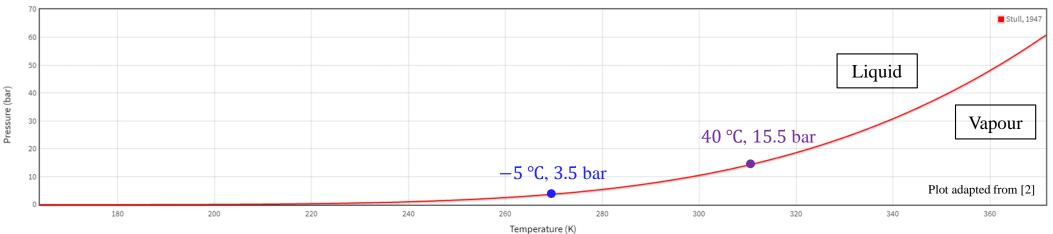


- Workspace exposure limits for anhydrous ammonia [3, 4]
  - o Long-Term (8 hours TWA): 25 ppm
  - Short-Term (15 minutes): 35 ppm
- Liquid to vapour expansion ratio for ammonia at room temperature: ~1:108 by volume [2]
- Recommended outdoor storage as per BCGA Code of Practice 18 [5]
- Need a liquid ammonia delivery system from outdoor storage to indoor test facilities



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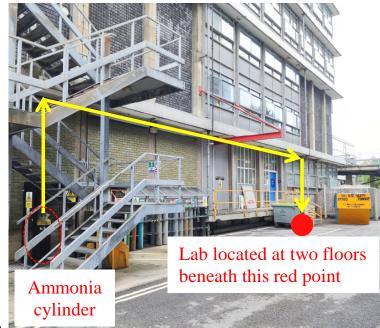
Ammonia cylinder from BOC



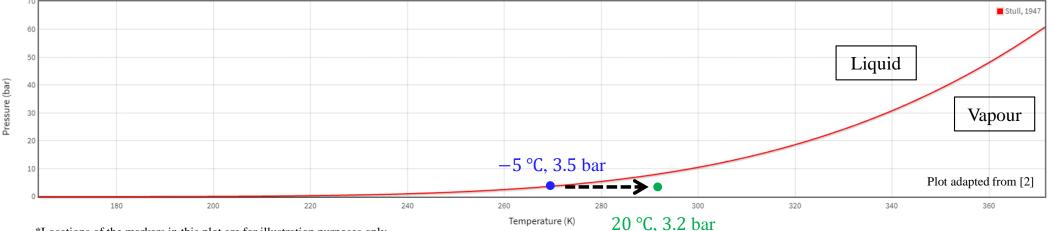
\*Locations of the markers in this plot are for illustration purposes only.

## **Ammonia delivery challenges**

- Chose against portable cylinders to minimise leakage risks during movement and re-connection
- Permanent ammonia delivery pipework (about 60 metres)
- Cavitation and evaporation of ammonia in the pipework
  - Pressure drop due to frictional loss
  - Temperature increase due to indoor heating
- Complete evaporation at the lab end in cold weathers
- Possible solutions:
  - Thermal insulation of the pipework
  - Liquefy ammonia at the lab end (accumulator) [6, 7]
  - Incorporate a pressurising system at the supply end (pump)



†Engineering Science Building in central Oxford



\*Locations of the markers in this plot are for illustration purposes only.

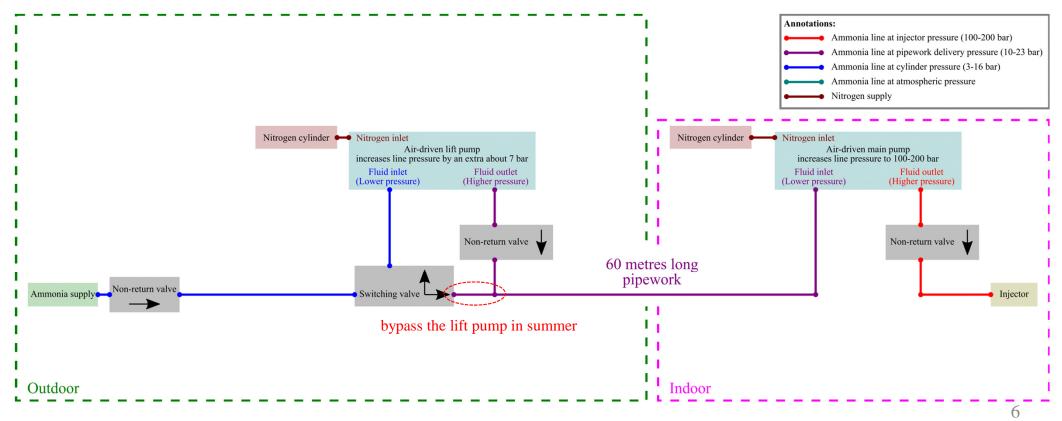
- Dual pump design
  - o Lift pump: increase pipework delivery pressure by an extra of 7 bar from cylinder pressure

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- Main pump: further increase liquid ammonia pressure to injector pressure
- Maintains a suitable in-line pressure for the lengthy pipework
- Offers a continuous liquid ammonia supply
- Limits amount of ammonia indoors (small bore pipework, ~5.5 mm in diameter)



# Safety features included

- Multiple venting points and options
  - Safety relief valve triggered at pre-set pressures: to atmosphere at an elevated location (emergency)

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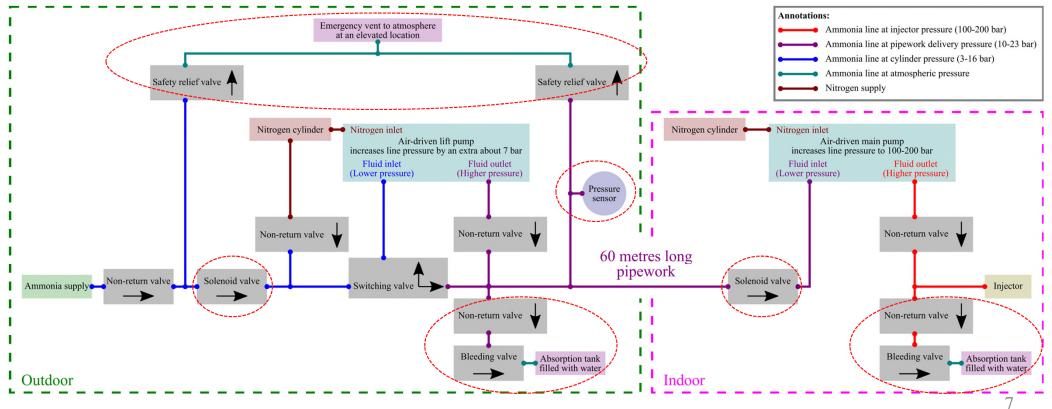
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- Manually operated bleeding valves: bubble through water (main)
- Ammonia-water solution sealed and disposed in UN certified containers
- Ammonia and in-line pressure sensors for leakage detection
- Solenoid valve power supply interlocks with leakage detection sensors



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### **Summary and future work**



- Ammonia supply as a saturation liquid only
  - Temperature dependent supply pressure
  - Lower supply pressure than required
- Toxicity considerations
  - o Outdoor cylinder storage
  - o Lengthy delivery pipework
- Challenges found: cavitation or even complete evaporation in pipework
- Solutions proposed: dual pump design
- Safety features
  - Emergency safety relief valve at pre-set pressures
  - Bubble waste ammonia into water and dispose resulting alkaline solutions
  - o Ammonia and in-line pressure sensors for leakage detection
  - Solenoid valve power supply interlocks with leakage detection sensors



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  - o Prof. Run Chen from Shanghai Jiao Tong University, China
  - o Dr. Abdelrahman Hegab from University of Nottingham, UK
  - o Mr. Ronan Pelé from University of Orléans, France

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Plot available at NIST database: <u>https://webbook.nist.gov/cgi/inchi?ID=C7664417&Mask=4&Type=ANTOINE&Plot=on#ANTOINE</u>

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